

## **Answer Key - Writing and Naming Formulas**

A.

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|--------------------|---------------------------------------|
| 1. $NaCl$          | sodium chloride                       |
| 2. $KClO_3$        | potassium chlorate                    |
| 3. $ZnSO_4$        | zinc sulphate                         |
| 4. $AlCl_3$        | aluminum chloride                     |
| 5. $Fe(ClO_3)_3$   | iron(III) chlorate OR ferric chlorate |
| 6. $(NH_4)_2SO_4$  | ammonium sulphate                     |
| 7. $Cu(NO_3)_2$    | copper(II) nitrate OR cupric nitrate  |
| 8. $Fe(CH_3COO)_2$ | iron (II) acetate OR ferrous acetate  |
| 9. $Ba(OH)_2$      | barium hydroxide                      |
| 10. $H_2SO_4$      | hydrogen sulphate or sulphuric acid   |
| 11. $H_2O$         | water                                 |

B.

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|------------------|-----------------|--------------------|
| 1. $NF_3$        | 8. $Sr(ClO)_2$  | 15. $Mg_3(Po_4)_2$ |
| 2. $NH_{3(g)}$   | 9. $MgCO_3$     | 16. $P_2O_5$       |
| 3. $AlBr_3$      | 10. $Ag_2CrO_4$ | 17. $Ca(HSO_4)_2$  |
| 4. $Sn(CrO_4)_2$ | 11. $Li_2SO_4$  | 18. $MnO$          |
| 5. $Ag_3PO_4$    | 12. $N_2O$      | 19. $Na_2O$        |
| 6. $Ca(OH)_2$    | 13. $H_{2(g)}$  | 20. $CaCl_2$       |
| 7. $Pb(NO_3)_2$  | 14. $NaHCO_3$   |                    |

C.

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|---------------------|-----------------------|
| 1. sodium carbonate | 6. carbon monoxide    |
| 2. aluminum oxide   | 7. hydrogen phosphate |
| 3. calcium nitrite  | 8. sodium sulphite    |
| 4. ammonium oxide   | 9. ammonium ion       |
| 5. neon gas         | 10. silver chlorite   |